# ECUACIONES QUÍMICAS

$MnO_2 + \mathbf{4}\ HCl \longrightarrow Cl_2 + MnCl_2 + \mathbf{2}\ H_2O$

$\mathbf{6}\ C + Al_2O_3 \longrightarrow Al_2C_3 + \mathbf{3}\ CO$

$\mathbf{2}\ CO_2 + Ca(OH)_2 \longrightarrow Ca(HCO_3)_2$

$\mathbf{2}\ H_2S + \mathbf{3}\ O_2 \longrightarrow \mathbf{2}\ SO_2 + \mathbf{2}\ H_2O$

$CH_4 + \mathbf{2}\ O_2 \longrightarrow CO_2 + \mathbf{2}\ H_2O$

$C_3H_8 + \mathbf{5}\ O_2 \longrightarrow \mathbf{3}\ CO_2 + \mathbf{4}\ H_2O$

$\mathbf{2}\ Na + Cl_2 \longrightarrow \mathbf{2}\ NaCl$

$\mathbf{4}\ Fe + \mathbf{3}\ O_2 \longrightarrow \mathbf{2}\ Fe_2O_3$

$C + O_2 \longrightarrow CO_2$

$\mathbf{4}\ FeS + \mathbf{7}\ O_2 \longrightarrow \mathbf{2}\ Fe_2O_3 + \mathbf{4}\ SO_2$

$\mathbf{2}\ C_4H_{10} + \mathbf{13}\ O_2 \longrightarrow \mathbf{8}\ CO_2 + \mathbf{10}\ H_2O$

$Na_2O + H_2O \longrightarrow \mathbf{2}\ NaOH$

$CaO + H_2O \longrightarrow Ca(OH)_2$

$CO_2 + H_2O \longrightarrow H_2CO_3$

$SO_3 + H_2O \longrightarrow H_2SO_4$

$N_2O_5 + H_2O \longrightarrow \mathbf{2}\ HNO_3$

$CaO + \mathbf{2}\ HCl \longrightarrow CaCl_2 + H_2O$

$\mathbf{2}\ NaOH + H_2S \longrightarrow Na_2S + \mathbf{2}\ H_2O$

$C_3H_8 + \mathbf{5}\ O_2 \longrightarrow \mathbf{3}\ CO_2 + \mathbf{4}\ H_2O$

$C_6H_{12}O_6 + \mathbf{6}\ O_2 \longrightarrow \mathbf{6}\ CO_2 + \mathbf{6}\ H_2O$

$\mathbf{2}\ HCl + Ca\ (OH)_2 \longrightarrow CaCl_2 + H_2O$

$\mathbf{3}\ H_2\ SO_4 + \mathbf{2}\ Fe\ (OH)_3 \longrightarrow Fe_2\ (SO_4)_3 + \mathbf{6}\ H_2O$

$\mathbf{4}\ HCl + O_2 \longrightarrow \mathbf{2}\ Cl_2 + \mathbf{2}\ H_2O$

$\mathbf{2}\ Na_3PO_4 + \mathbf{3}\ Ba(NO_3)_2 \longrightarrow Ba_3(PO_4)_2 + \mathbf{6}\ NaNO_3$

$\mathbf{2}\ C_2H_6O_{(l)} + \mathbf{7}\ O_{2\ (g)} \longrightarrow \mathbf{4}\ CO_{2\ (g)} + \mathbf{6}\ H_2O_{(l)}$